# The impact of disinfectants on residue formation on endoscope surfaces

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# **Topics**

- Objectives
- Options for active substances
- Impact of residue formation depending on active substance
- Conclusions

# Objectives

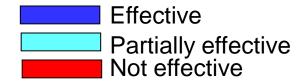
Focus on residue formation by interaction between disinfecting agents and proteins

Two questions will be discussed

- A) Are there large molecules formed?
- B) Does the disinfecting agent support the link between large molecule and the endoscope surface fixation?

### **Options for Active Substances**

Range of effectiveness of various disinfecting agents



	Halogens	Peroxides	Aldehydes	Alcohols	Phenols	QAV/Biguan	Amines	Acids
Gram neg. bact.								
Gram pos. bact.								
Mycobact.								
Bacteria spores								
Yeasts								
Molds								
Unencap. viruses								
Encap. viruses								

"Reactive effective agents" Non-reactive effective agents

### **Options for Active Substances**

Active Substances used for endoscope disinfection:

### Aldehydes:

- ✓ Glutaraldehyde
- √ o-Phthalaldehyde (OPA)

### Oxidising Substances:

- ✓ Chlorine dioxide
- ✓ Hypochlorous acid
- √ Hydrogen peroxide
- ✓ Buffered peracetic acid

### **Options for Active Substances**

Active Substances used for endoscope disinfection:

### Aldehydes:

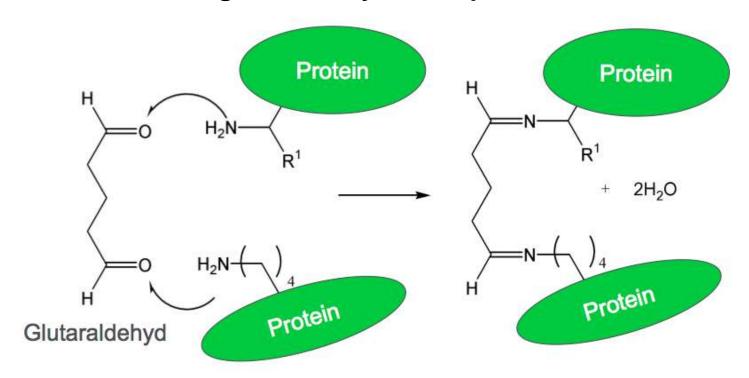
- ✓ Glutaraldehyde
- √ o-Phthalaldehyde (OPA)

### Oxidising Substances:

- ✓ Chlorine dioxide
- ✓ Hypochlorous acid
- √ Hydrogen peroxide
- ✓ Buffered peracetic acid

# Glutaraldehyde Formation of large molecules

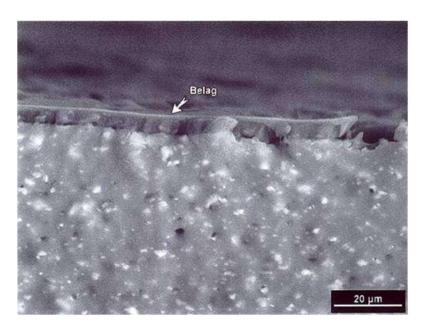
Reactions between glutaraldehyde and proteins:

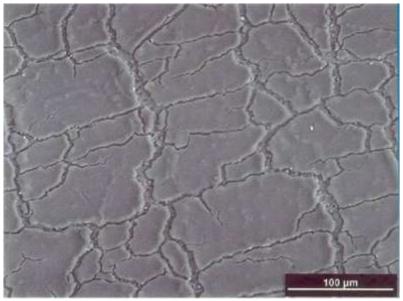


- Glutaraldehyde is a protein cross-linker.
- Large molecules are formed.

# Glutaraldehyde Fixation of large molecules

➤ Deposits on the outer surfaces of insertion tube are observed.





Magnification: 1000

Magnification: 275

Source: M. Kamer, Dr. Weigert

# Glutaraldehyde Fixation of large molecules

Deposits on the other surfaces of insertion tube are observed.

Residues in channels are observed as well.

#### Reason:

Insufficient protein removal after use and before disinfection with glutaraldehyde?

#### **Question:**

Why is the residues limited on the area of insertion tube coming in contact with the patient?

# Glutaraldehyde Fixation of large molecules

### **Explanation:**

- ➤ Glutaraldehyde adsorbs in the disinfection phase in small amount on the plastic surfaces and will not be complete removed by the following final rinsing.
- ➤ Absorbed glutaraldehyde reacts with proteins during the endoscope contact with the patient.
- ➤ Formed large molecules can not totally removed in the following reprocessing procedure.
- Residue layers are built-up in several reprocessing cycles.

# Glutaraldehyde Summary

- Consider residue formation by fixation of large glutaraldehyde-protein molecules during endoscope use.
- > Strong focus on
  - improved final rinsing
  - Ly mechanical support in the cleaning phase due to difficult to remove larger molecules

# o-Phthalaldehyde (OPA)

### Formation of large molecules:

- > OPA reacts with proteins
- Probability to form large molecules is much lower than glutaraldehyde

### Fixation of large molecules:

- Likelihood low
- Less experience
- Should be investigated

# Oxidising Acids Impact of pH-value

**Example: Hydrochloric acid** 



**Acid** 



Salt



### **Oxidising acids:**

Hypochlorous acid



Buffered hypochlorous acid



Sodium hypochlorite

Peracetic acid



Buffered peracetic acid



Sodium peracetate

# Oxidising Acids Impact of pH-value

**Example: Hydrochloric acid** 



**Acid** 



Salt



### Oxidising acids used in endoscope disinfection:

Hypochlorous acid



Buffered hypochlorous acid



Sodium hypochlorite

Peracetic acid



Buffered peracetic acid



Sodium peracetate

### Hypochlorous acid

### Formation of large molecules:

- > Acid related protein coagulation is possible.
- Intermolecular reactions have been observed with milk proteins.

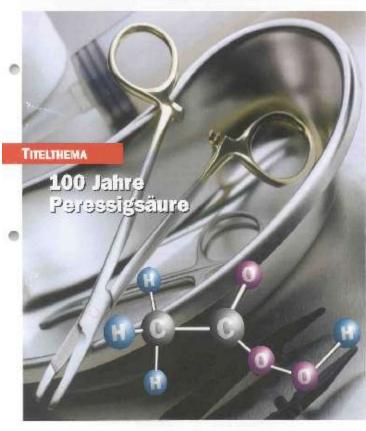
### Fixation of large molecules:

#### Likelihood low because of:

- low concentration of active agent
- fast reaction between hypchlorous acid and proteins
- device detect the lowest accepted concentration
- decomposition of active agent no adsorption

# Peracetic acid History





- 1902 First publication on the microbiological effect of peracetic acid by Frier and Novy
- 1949 Comparison of 23 antimicrobial effective agents
  - Peracetic acid is the most effective substance
- 1955 First use of peracetic acid in raising experimental animals free of germs
- 1960 70 Ground-laying experiments on stability, analysis and material compatibility in work groups in Erfurt and Prague
- 1970 Introduction of first peracetate-based antimicrobial cleaner for surgical instruments to the market

## Peracetic acid and its salts Applications in the medical field

Antimicrobial cleaner for surgical instruments

pH-value: alkaline

Disinfection of haemodialysis devices

pH-value: acid

Chemo-thermal disinfection procedure for hospital linen tested by RKI

pH-value: alkaline

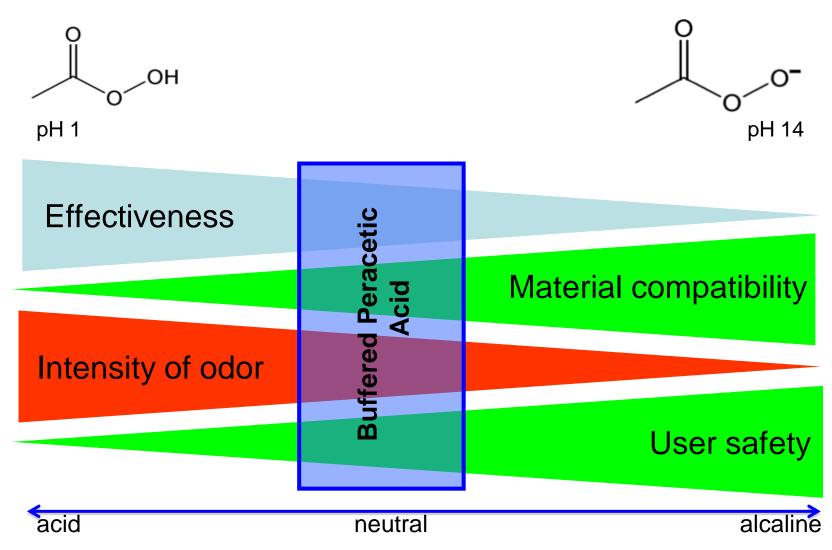
Hand and skin disinfection

pH-value: buffered

Disinfection of medical instruments, including flexible endoscopes

pH-value: buffered

# Peracetic acid and its salts pH-value impact



# Buffered peracetic acid Formation of large molecules

Free-radical reaction with proteins

Decomposition of protein chains by peracetic acid

Kerkaert B, et al. J Agric Food Chem (2011) 59: 907-914

# Buffered peracetic acid Fixation of large molecules

- ➤ One lab study with artificial blood described the fixation of large protein molecules (fibrin) on stainless steel plates.
- Fixation of fibrin could not observed on synthetic surfaces in other lab studies
- Field studies and practical experience in endoscope reprocessing show no residue formation on endoscope surfaces

## Buffered peracetic acid Remove of glutaraldehyde-protein deposits

Endoscope channels after routine reprocessing with glutaraldehyde followed by cycles with buffered peracetic acid:



After routine disinfection with glutaraldehyde



After 30 cycles with peracetic acid and no mechanical support



After 30 cycles with peracetic acid and with mechanical support

Meyer B. HygMed 2004; 29: 106-109 Tucker RC, et al., ASAIO J 1996; 42: 306-313

### Buffered peracetic acid Remove of glutaraldehyde-protein deposits

### **Practical experience:**

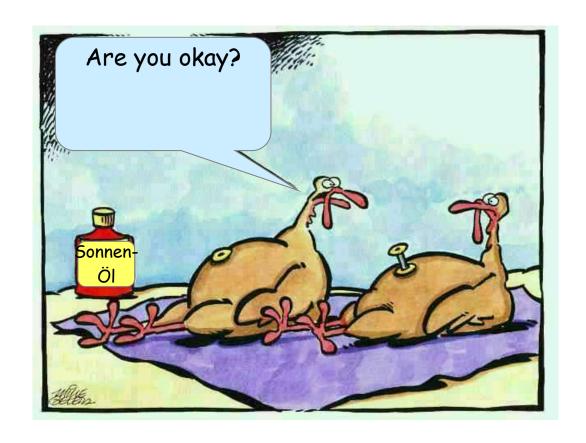
- Deposit remove on outer surfaces after couple of reprocessing cycles
  - Ly Brightening the ring markers
- Stiffness of brushing during the transition period
  - Ly Deposits are removed

# Buffered peracetic acid Summary

- Because of steric hindrance, intermolecular reactions plays a minor role in interaction with proteins.
- > Formation of bigger molecules are not expected.
- ➤ Even though lab study show fixation of large molecules on stainless steel plates, comparable effects on synthetic materials are not observed.
  - ho residue formation under reprocessing conditions in the field
- ➤ Glutaraldehyde/Protein deposits are removed by repeated disinfection with buffered peracetic acid.

### Conclusions

- Glutaraldehyde cross-link proteins
  - Risk of residue formation should be considered.
  - Ly enhanced rinsing, brushing and cleaning is required.
- Minor probability for protein cross-link and fixation of formed molecules in case of o-phthalaldehyd (OPA), hypochlorous acid and buffered peracetic acid
- ➤ Glutaraldehyde/Protein deposits can be removed by repeated treatment/disinfection with buffered peracetic acid



Thank you for your attention!